



HTC Project 3b

High Temperature Corrosion in systems for heat storage (corrosion in melts for concentrated thermal solar power)

Presented by LG. Johansson, E. Hamdi and P. Dömstedt

Project Organization

- Project leader: LG Johansson, Chalmers
- Companies:





- Researchers in HTC: LG Johansson, JE Svensson, C Geers, P Szakalos
- PhD students: E. Hamdi, P. Dömstedt





Project plan

- Project Goals:
- Investigate corrosion processes relevant to Concentating Solar Power (CSP) plants in order to generate new ideas and strategies for solving the serious materials challenges facing this industry. The project deals with the interaction between melts used for heat transport (HTF) and heat storage and focuses on the mechanisms of corrosion attack



Project plan

The project consists of two parts:

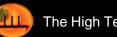
- Corrosion in carbonate and chloride melts AND
- Corrosion in an Al-Si eutectic melt



Background

State-of-the-art CSP plants that generate power in a steam turbine presently use a eutectic mixture of sodium nitrate and potassium nitrate (= "solar salt"). Because that mixture decomposes spontaneously > 600°C, the drive towards higher power efficiencies and higher temperatures (>700°C) has meant that more thermally stable HTF's, based on alkali carbonates, chlorides, fluorides or liquid metals are now being considered. However, today's plants are constructed of materials that are unable to withstand corrosion at the higher working temperature in the presence of the "new" HTF's.

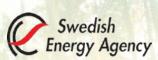
















High Temperature Corrosion of alloys in Molten Salts for Heat **Storage**

Part I: Molten Salts

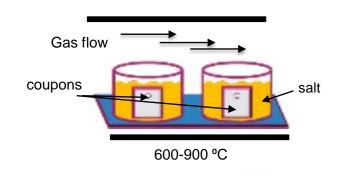




EXPERIMENTAL SET UP









- O Partial immersion of coupons
- O Refilling is needed for long exposures
- O After Exposure "for carbonates salt":

 coupons are washed out in warm water (60 °C) with ultrasonic treatment
- O Gravimetric investigations, not possible!





"DREAM" SET UP!

- O Fully immersion of coupons
- O Carbonate blend freezes ~ 450 °C
 - If the salt freezes while the coupon is still immersed
 - Salt needs to be washed out, 60 °C with ultrasonic
 - Hard to evaluate highly biased surface (leached)
 - Weigh the coupons after the exposures!





Longer cylinders, Longer crucibles?

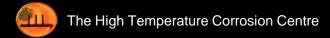




Temperature 750-1000 °C Protect the cylinders!



Two cylinders,
More crucibles
6*2= 12 coupons at one exposure,
same conditions



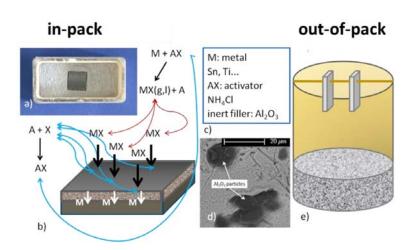
NEW SET UP,

CHALLENGES & SOLUTIONS

1. Cylinder, 253 MA

At high temperature exposures, may not survive?!

Aluminum diffusion coated furnace cylinders





C. Geers, "Inhibition of coking and metal dusting on conventional alloys by using a nickel-tin intermetallic coating," 2013.





NEW SET UP,

CHALLENGES & SOLUTIONS

- ✓ Longer crucibles, longer exposures without refilling.
- ✓ Coupons are totally immersed in the salt.
- ✓ Easier removal of salt.

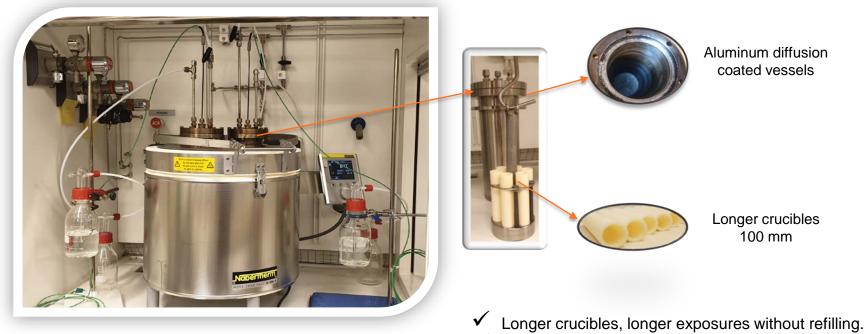








EXPERIMENTAL SET UP



New set up

- Coupons are totally immersed in the salt.
- Easier removal of salt.



The H

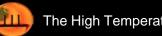
EXPERIMENTAL SET UP (CONT.)

Molten Salt	Nitrates (60 wt% NaNO ₃ - 40 wt% KNO ₃)	Carbonates (33 wt% Li ₂ CO ₃ -32 wt% Na ₂ CO ₃ - 34 wt% K ₂ CO ₃)	Chlorides (65 wt% KCI- 32% MgCl ₂)
Gas	Air	CO_2	Argon
Temperature Range °C	550-680	750-900	800-1000

- ✓ Ar, CO₂ connected
- ✓ Temperature Range ~ 1000 °C
- ✓ Long exposures without refilling 10-13 g of salt instead of ~1,7 g
- ✓ Role of oxygen and water as crucial impurities
- ✓ Sample dimensions



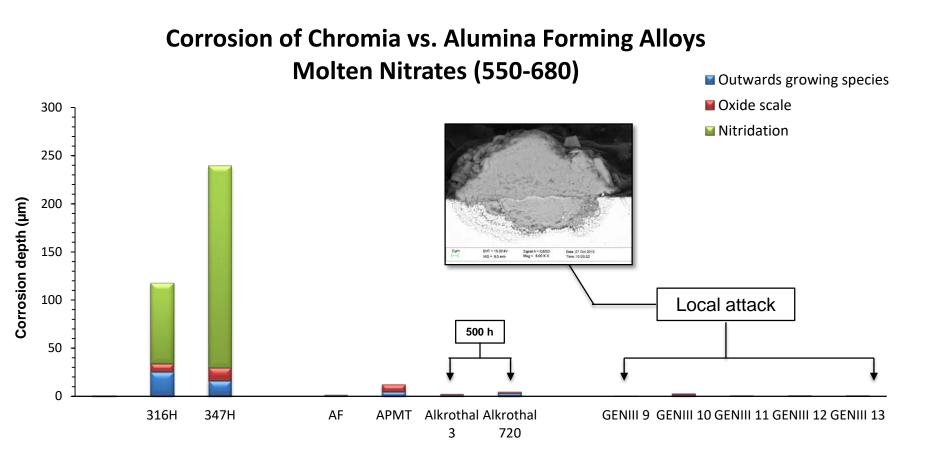




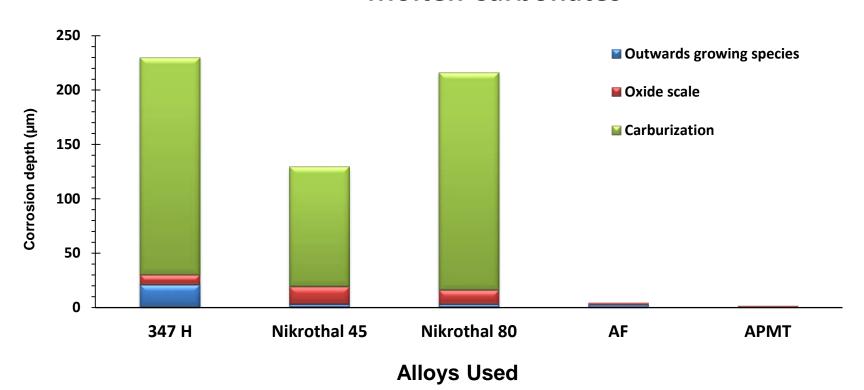
NEXT SET UP!!

Planning to extend our work Building new set up! Suggestions?!!





Corrosion of Chromia vs. Alumina Former Alloys Molten Carbonates





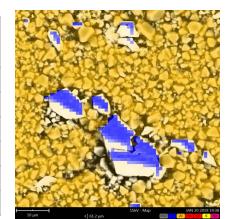
NEW EXPOSURES AT 900 °C

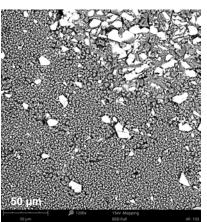
Kanthal AF

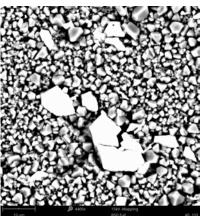
- Macroscopic analysis
- Optical imaging
- SEM/EDX on BIB prepared cross sections
- Structural characterization

Element Number	Element Symbol	Element Name	Atomic Conc.	Weight Conc.
8	0	Oxygen	66.09	50.91
13	<mark>Al</mark>	Aluminium	<mark>29.84</mark>	38.77
26	<mark>Fe</mark>	Iron	<mark>2.82</mark>	7.59
24	<mark>Cr</mark>	Chromium	<mark>0.83</mark>	2.07
19	К	Potassium	0.25	0.47
11	Na	Sodium	0.18	0.19











ACKNOWLEDGMENT



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HTC 3b - Corrosion resistant materials in molten Al-Si alloy

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Image Source:

Azelio Technology

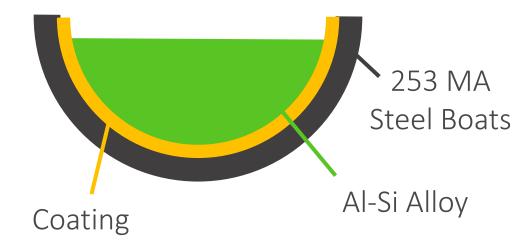
Al-Si Alloy Corrosion test

- Stirling based CSP with TES
 - Al-Si alloy as PCM
 - Na as HTF
- 253 MA steel with corrosion protective coating against Al-Si alloy

- Cyclic exposure from 540-600°C
 - T_{melt}(Al-Si) **577°C**
- COSTA tube furnace (KIT)
- 253 MA containers
 - Coated on the inside
 - Filled with AL-Si alloy

Al-Si Alloy Exposures

Tube Furnace Gas In Samples Gas Out Samples



Aluminium titanate Al₂TiO₅

Other coatings

- Good corrosion protection against molten
 Al
- Low wettability with molten metals
- Properties of Al₂TiO₅ coatings:
 - Good shock resistance
 - Stable up to over 1000°C
 - High **chemical** and **thermal** stability
 - Low thermal expansion
 - Low thermal conductivity

Collaboration with Nanoker (Spain)

Dopants altering coating properties (thermal expansion)

- Titanium Diboride (TiB₂)
 - high hardness and wear resistance
 - high thermal expansion and conductivity
- Boronising (FeB, Fe₂B)
 - high hardness and wear resistance
 - high thermal expansion and conductivity
- Yttria Stabilised Zr (YSZ)
 - High chemical and thermal stability
 - Intermediate thermal expansion

Steel/Coating Properties

Material/Coating	Therm. Expan. Coef. (10 ⁻⁶ 1/K)	Thermal Conductivity (W/mK)
FeB (Boronising)	~ 20	~ 12
Austenitic Steels	~ 15	~ 16-22
Fe ₂ B (Boronising)	~ 10	~ 20
Ferritic/Martensitic Steels	~ 10	~ 25-30
TiB ₂	~ 7	~ 25
YSZ	~ 5	~ 1
Al_2TiO_5	~ 1	~ 1.5

References '

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