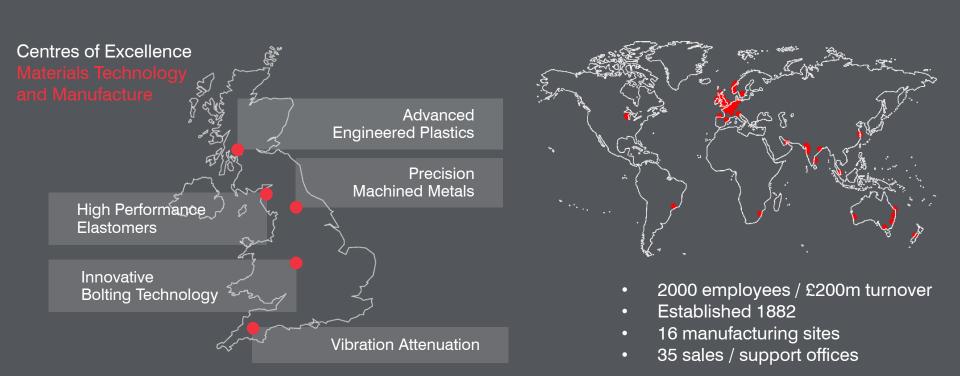
# Compounding and Manufacturing Elastomers for Nuclear Applications

Polymers in nuclear applications 2019 Andrew Douglas FIMMM

#### James Walker Overview



#### **James Walker**

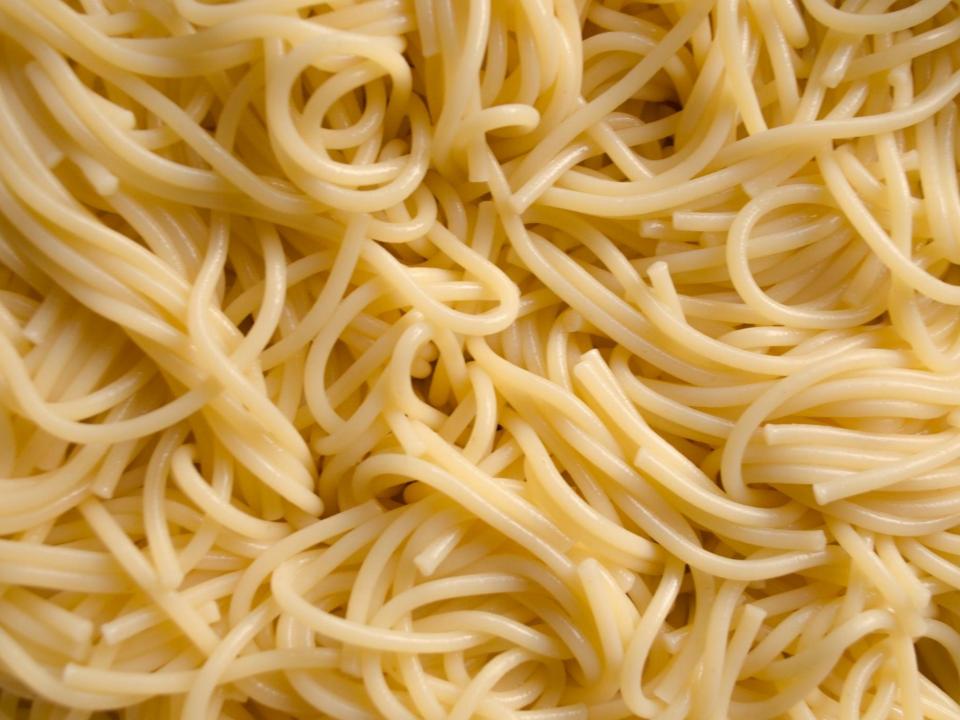


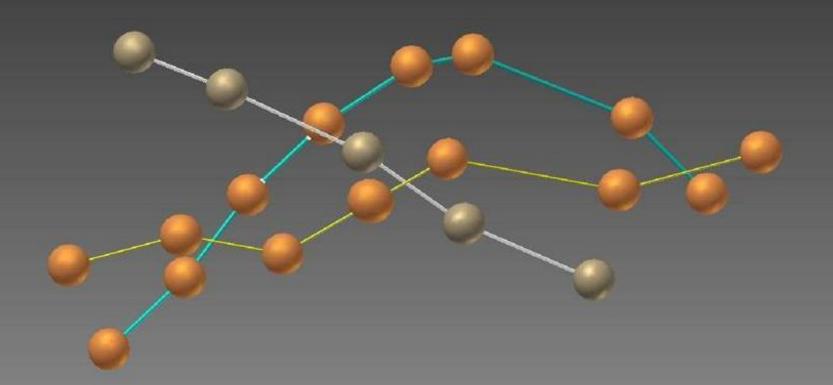
## What is an Elastomer? The concept of polymers

## Atoms

# Molecules

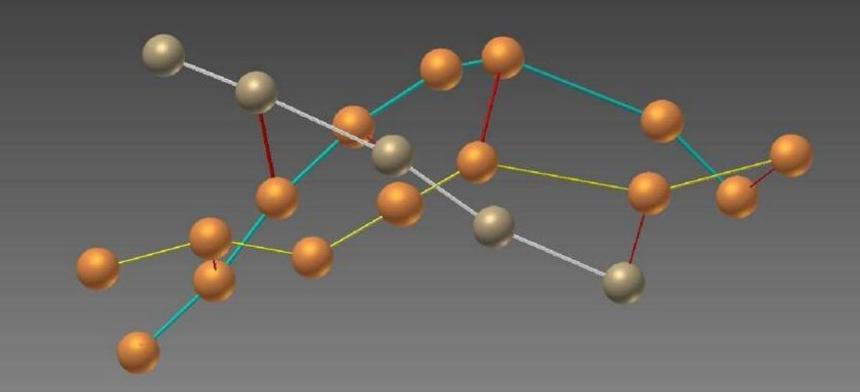
# Polymers



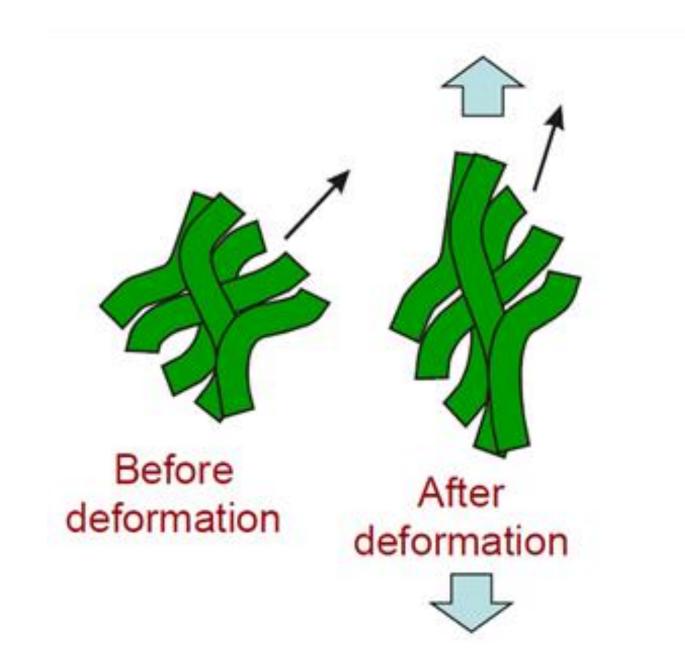








Polymer structure with **Crosslinks** 





#### IMPLEMENTS USED BY BRAZILIAN RUBBER COLLECTORS

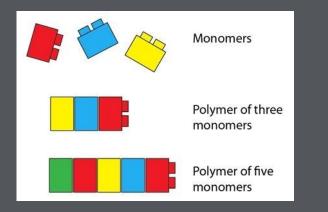
A. "MACHADINHA' SMALL ANE) FOR TAPPING. E. COLLECTING CUP. C. COLLECTING VESSEL INTO WHICH CONTENTS OF CUPS ARE POWRED. E. SAUCER BY MEANS OF WHICH LATEX IS POWRED OVER PADDLE F DURING SMOKING PROCESS. THE PADDLE F (TO THE LEFT) IS SHOWN COVERED WITH A LAVER OF RUBBER

# The history of natural rubber

The development of Synthetic rubbers and Compounding

#### Designing synthetic polymers...

#### Monomers & Polymers



#### Attributes of monomer 'building blocks'

Strength of bonds within molecules and between them

Highly polar gives resistance to swelling by oils/fuels

Flexibility at low temperatures

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# Synthetic Polymers

# How long will it last?

'All of the disputed territories contain valuable minerals, and some of them yield important vegetable products such as rubber which in colder climates it is necessary to synthesize by comparatively expensive methods'

Nineteen Eighty-Four, George Orwell, first published June 1949

Synthetic elastomers haven't actually been around for that long so

100 year life predictions come with caveats...

# **GENERAL ELASTOMER CLASSIFICATION**

#### Natural

#### Chloroprene

Ethylene-Propylene General Purpose Low oil resistance General Purpose Oil Resistant

Aflas®

Polyurethane

Nitrile

High Performance

Fluorocarbon

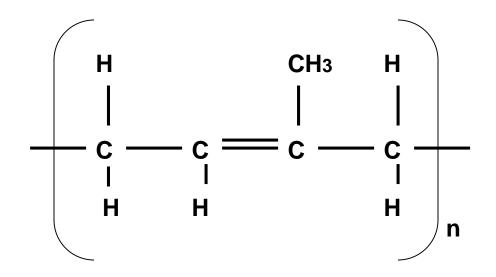
Silicone

HNBR

Acrylic

## Natural Rubber (NR)

Cis- 1,4- polyisoprene



<u>Advantages</u>: High resilience; high tensile/tear; good abrasion resistance; low cost.

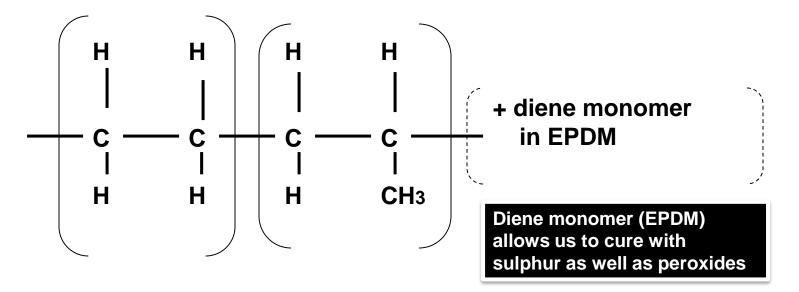
**Disadvantages**: Poor oil resistance; poor weathering resistance.

Reversion.

**Typical temperature range:** -50°C to +100°C

### **Ethylene-Propylene Rubber (EPM, EPDM)**

Dipolymer of ethylene and propylene (EPM), or Termonomer of ethylene-propylene with a diene monomer



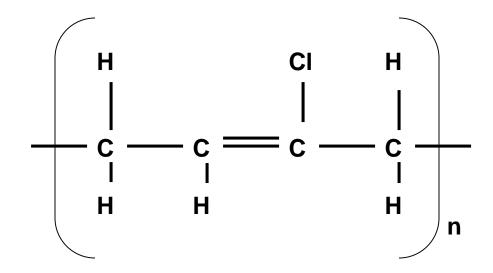
**Advantages:** Excellent ozone/weathering resistance. Excellent radiation, hot water and steam resistance (peroxide grades); good resistance to inorganic and polar organic chemicals.

**Disadvantages:** Low resistance to hydrocarbons.

Typical temperature range: -45°C to +150°C (-49°F to +300°F).

## Polychloroprene (Neoprene) Rubber (CR)

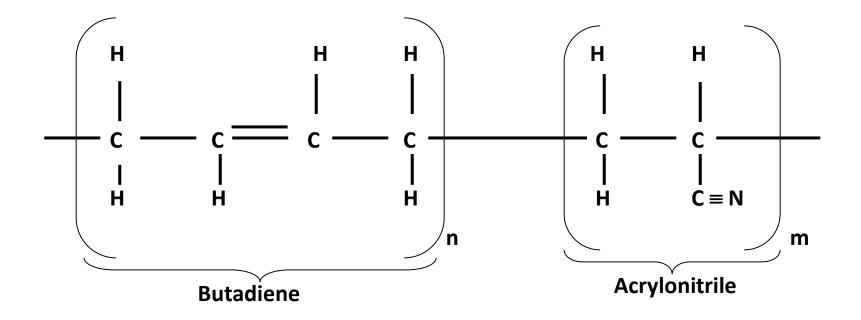
2-chlorobutadiene



<u>Advantages:</u> Good weather and ozone resistance; good mechanicals; resistance to many Freons; some oil resistance; low cost. <u>Disadvantages:</u> Only moderate oil resistance; limited temperature resistance. <u>Typical temperature range:</u> -40°C to +120°C ( -40°F to +250°F)

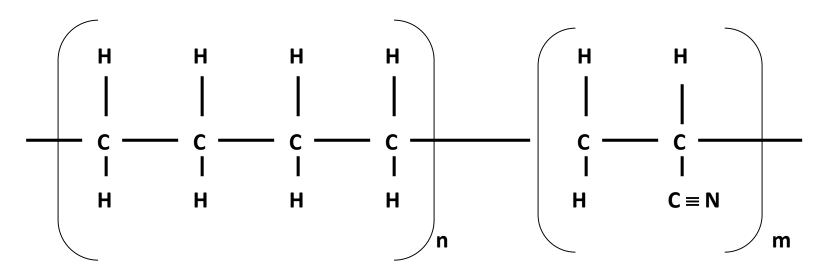
## Nitrile Rubber (NBR)

Acrylonitrile-butadiene



**Typical temperature range:** -50°C to +120°C (dependent on ACN content!)

### Hydrogenated Nitrile Rubber (HNBR/HSN)

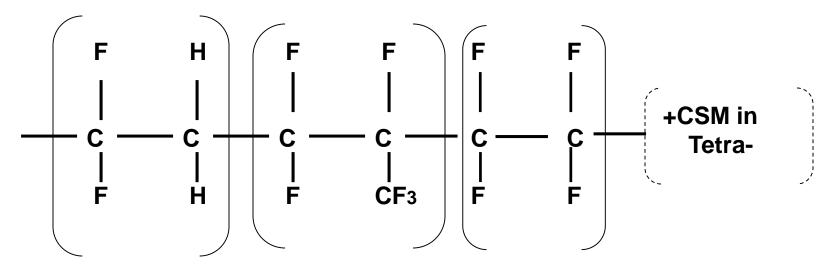


Made from NBR by hydrogenation reaction

**Typical temperature range:** -40°C to +160°C (dependent on ACN content!)

### Fluoroelastomers (FKM/FPM)

Eg., 'Viton<sup>®</sup>', 'Fluorel<sup>®</sup>', 'Tecnoflon<sup>®</sup>, DAI-El ™

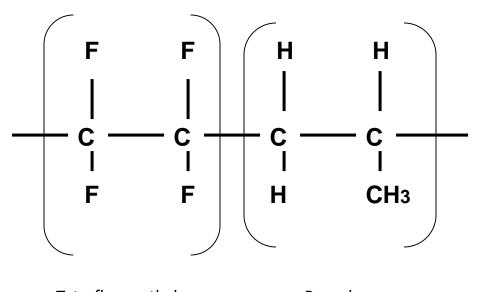


Vinylidene fluoride, VDF (VF<sub>2</sub>) Hexafluorpropylene, HFP Tetrafluoroethylene, TFE Cure site monomer, CSM

**Dipolymers**, VDF and HFP, ~66%F ('A' Types) **Terpolymers\*,** VDF, HFP, TFE (CSM) ~68%F to ~70%F FKM ('B' types at 68%F and 'F'types at 70%F). **Typical temperature range:** -45°C to +200°C

## Aflas® (TFE-P / FEPM)

Tetrafluoroethylene-propylene



#### **Speciality FKM**

Tetrafluoroethylene Propylene

<u>Advantages:</u> Excellent ozone/weathering resistance; good heat resistance; excellent resistance to steam and radiation; good overall chemical resistance. <u>Disadvantages:</u> High compression set; high Tg; poor resistance to aromatics <u>Typical temperature range:</u> 0°C to +200°C

# **RELATIVE BASE POLYMER COSTS...**







# Protective system

## Cure system

What **GOES in** to an elastomer?

#### **COMPOUND INGREDIENTS**

POLYMERS **CURATIVES ACCELERATORS ACTIVATORS** RETARDERS **CO-AGENTS FILLERS** PLASTICISERS **EXTENDERS PROTECTANTS** 10

1

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**BONDING PROMOTERS** 11 **PROCESS AIDS** 12 **TACKIFIERS** 13 **DESICCANTS** 14 PIGMENTS 15 **BLOWING AGENTS** 16 **PEPTISERS** 17 **FLAME RETARDANTS** 18 ODOURANTS 19 **COUPLING AGENTS** 20

	High Quality	Low Quality
34% ACN NBR	100	100
Zinc oxide	5	5
Stearic acid	1	1
Sulphur	1	2
Accelerators	4	4
Ester plasticiser	7	-
$AO_2/AO_3$	2	-
Carbon black N550	70	20
Oil extender	-	50
Chalk Dust	-	200
Talcum Powder	-	50
White filler	-	50
Wax	-	5
Total PHR	190	487

	High Quality	Low Quality
Cure	6 minutes at 180°C	
Hardness (IRHD)	78	72
Tensile Strength (MPa)	19.3	2.4
E @ B (%)	300	100
Modulus at 25% strain (MPa)	1.3	0.8
Modulus at 50% strain (MPa)	2.2	0.9
Modulus at 100% strain (MPa)	4.8	1.0
Tear strength (N/mm)	42	9
Compression Set, 24hr @ 70°C (%)	7	40
Specific gravity	1.2	1.6

# One piece of rubber looks pretty much like another....

**James Walker** 

# Elastomer behaviour at **IOW** temperatures

- Decrease in energy
- Polymer chains lose flexibility
- 'Leathery phase' before taking on the properties of glass
- Properties vary depending on the polymer type/formulation

# Elastomer behaviour at high temperatures

- Long-term resistance influenced by strength of bonds between atoms
- Stronger bonds permit higher operating temperatures
- Mechanical properties will reduce

# Elastomer behaviour in **CONTACT MEDI**

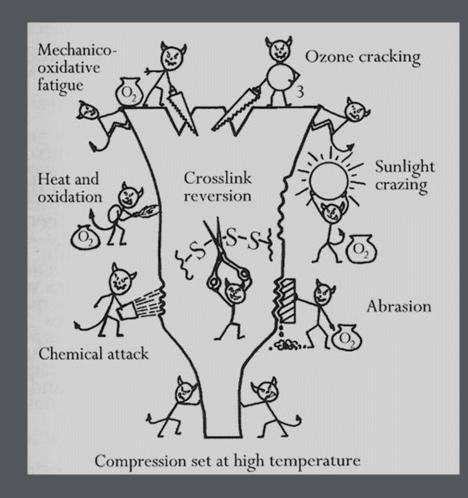
## Swelling

- Elastomer will absorb the contact media and swell
- Physical effect, which is to some extent reversible

### **Chemical Attack**

- Polymer chain is altered / attacked by the contact media
- Irreversible

# Can occur at the same time



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**Environmental Qualification** 

#### **Fixed formulations – fantastic**

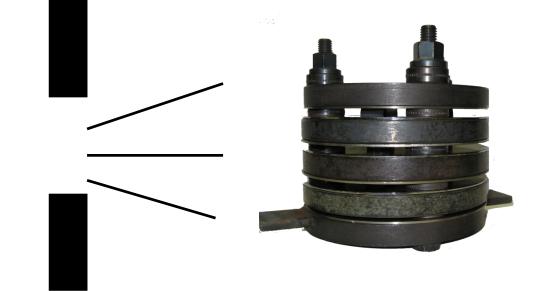
But...

Legacy materials – availability of raw ingredients

**James Walker** 

# Shieldseal® 661, 663 and 664 Elastomers

Radiation testing carried out by Wood Group



# <sup>60</sup> Co

- Exposure to gamma radiation
- 1kGyhr<sup>-1</sup> up to 1000 kGy
- Followed by further 600 kGy
- Aged and unaged samples
- Intended to replicate LOCA broadly based on IEEE 383 / IEC 60780 standards

# Accelerated Thermal Ageing

Potential impact of variations in Activation Energy



Victoria Smith Wood Group EQSA Seminar Sept 2019

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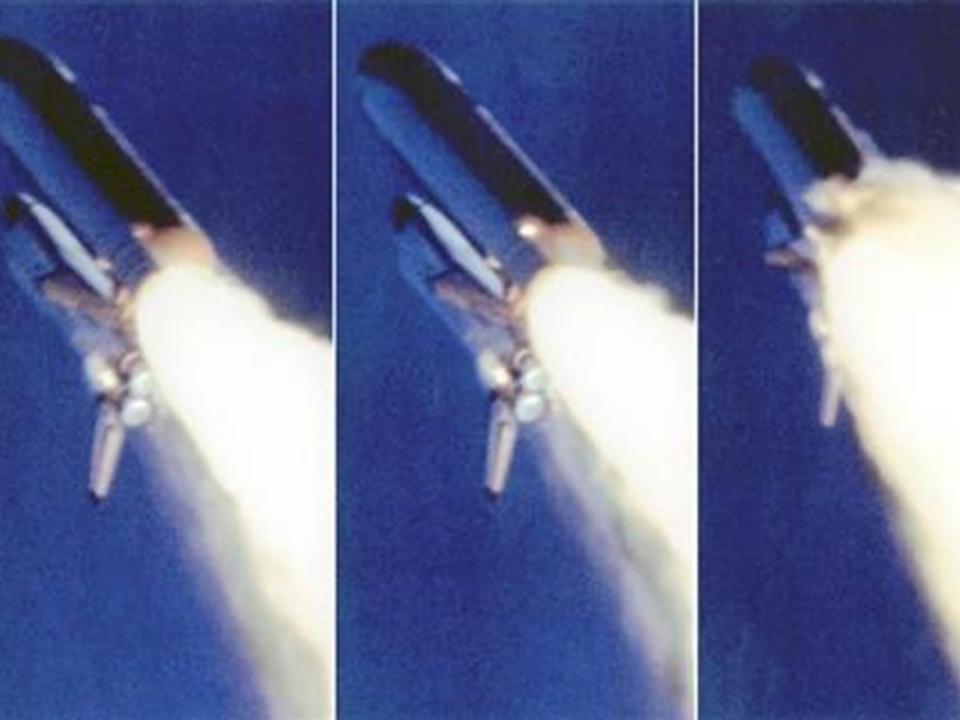


#### Martin Baker – Ejector seats – O

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# Thank you!

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